



Energy transfers in a system

- ① Understand that energy can be transferred usefully, stored or dissipated.
- ② Explain how to reduce unwanted energy transfers.

Energy transfers in a system



Energy can be usefully transferred, stored or dissipated. Energy cannot be created or destroyed.

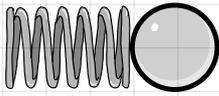
Open and closed systems



A **closed** system is when there is **no net change** in the total **energy**. The types of energy within the system may change.



An **open** system is when energy can be transferred between the system and its surroundings. The total **energy** in the system can **increase** or **decrease**.



An example of a closed system is a pinball machine. When the spring is released the elastic potential energy stored is converted to kinetic energy in the ball.

An example of an open system is a cup of coffee. As steam escapes from the coffee, thermal energy is transferred to the surroundings. Over time the coffee becomes cold (total energy is decreased).

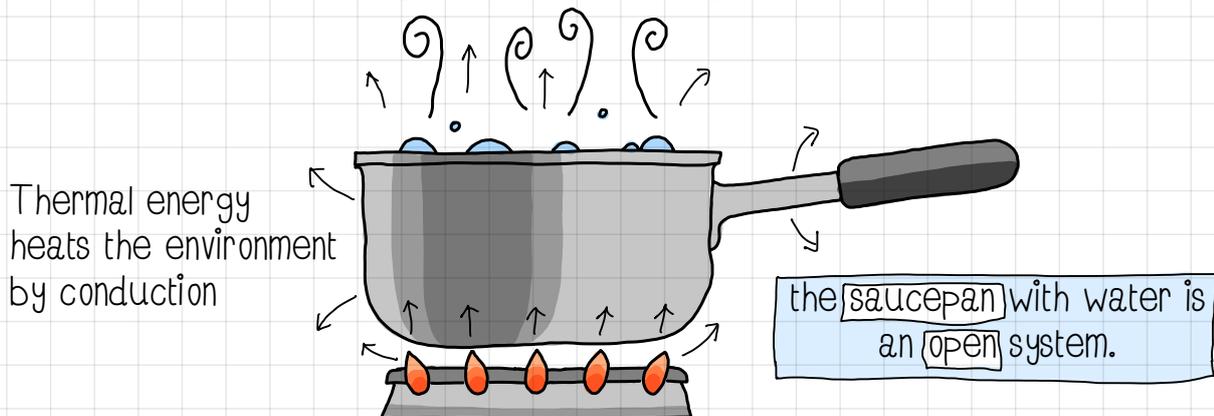


Dissipated energy



When energy is dissipated, it is transferred to a **less useful** store (usually thermal energy). This energy is often described as being wasted.

Steam dissipates thermal energy to the surroundings



Thermal energy is added to the saucepan and water, increasing the total useful energy

Energy transfers in a system...

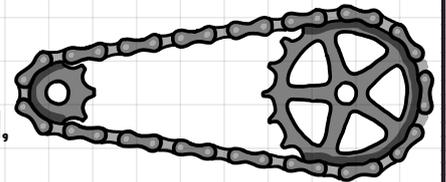
Reducing unwanted energy transfers



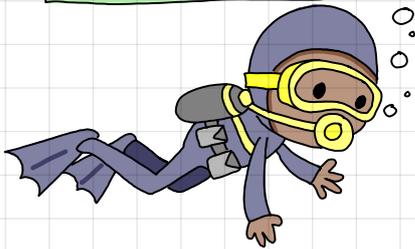
Unwanted energy transfers can be reduced through lubrication and thermal insulation.

Lubrication

Friction between the gear and the chain causes energy to be dissipated to the surroundings as thermal energy. By **lubricating** the chain with oil, **friction** is **reduced** and less energy is dissipated



Thermal insulation



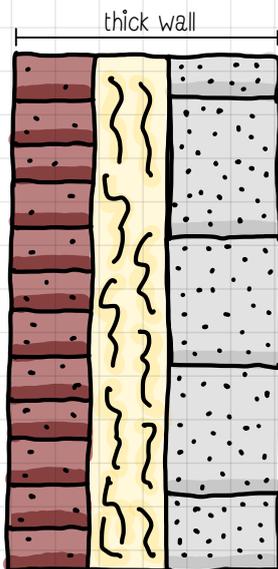
Heat from an object can be quickly dissipated to the environment. Insulation can be used to **reduce** the **transfer** of thermal energy. Examples of thermal insulation include foam coffee cups, clothing and loft insulation

Thermal conductivity



The higher the thermal conductivity of a material, the higher the rate of energy transfer by conduction across the material. Metals and other **conductors** have a **high** thermal conductivity. **Insulators** have a **low** thermal conductivity.

Cooling in buildings



The rate of cooling in buildings can be reduced by the use of **insulating** material in the walls which has a low thermal conductivity. **Thick walls** also reduce the cooling rate. **Double glazing** traps air between two panes of glass. Air has a low thermal conductivity. Materials with low thermal conductivity can also be used in the **loft** to reduce the rate of cooling.

